



E. Compound Semiconductors 분과

Room K
청옥II+III(6층)

2016년 2월 24일(수) 10:10-11:40

[WK2-E] III-V Device

좌장 : 민병규(한국전자통신연구원), 김해천(한국전자통신연구원)

High Performance $\text{In}_{0.7}\text{Ga}_{0.3}\text{As}$ MOSFETs with $\text{Al}_2\text{O}_3/\text{HfO}_2$

WK2-E-1
10:10-10:25

Seung Woo Son, Jin Su Kim, Hwal Kim, Jung Ho Park, Do-Kywn Kim, Jung-Hee Lee, and Dae-Hyun Kim

School of Electronics Engineering, Kyungpook National University

Oxidation Study on the (100), (110) and (111) Surfaces of InAs by ab-initio Calculations

WK2-E-2
10:25-10:40

In Won Yeu^{1,2}, Cheol Seong Hwang^{2,3}, and Jung-Hae Choi¹

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Improvement of Thermal Stability of Ni-InGaAs on Source and Drain by using Pd Interlayer for High Performance n-InGaAs MOSFET

WK2-E-3
10:40-10:55

Meng Li¹, Jeyoung Kim¹, Jungwoo Oh², and Hi-Deok Lee¹

¹Department of Electronics Engineering, Chungnam National University,

²School of Integrated Technology, Yonsei University

Universal Mobility Behavior in $\text{In}_{0.7}\text{Ga}_{0.3}\text{As}$ QW-MOSFETs

WK2-E-4
10:55-11:10

Jung Ho Park, Hwal Kim, Do-Kywn Kim, Jin Su Kim, Seung Woo Son, Jung-Hee Lee, and Dae-Hyun Kim

School of Electronics Engineering, Kyungpook National University

The Fabrication of InGaAs MOSFET with Y_2O_3 Gate Insulator

WK2-E-5
11:10-11:25

Seong Kwang Kim^{1,2}, Dae-Myeong Geum^{2,3}, Jungmin Lee¹, Min-Su Park², Jae-Phil Shim², Chang Zoo Kim⁴, Hyung-jun Kim², Jin-Dong Song², Won Jun Choi², Sung-Jin Choi¹, Dae Hwan Kim¹, SangHyeon Kim², and Dong Myong Kim¹

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Improvement of Interfacial-state Density (D_{it}) in High-k/ $\text{In}_{0.53}\text{Ga}_{0.47}\text{As}$ MOSCAPs by D_2 High-Pressure Annealing (HPA)

WK2-E-6
11:25-11:40

Jin Su Kim¹, Seung Heon Shin², Do-Kywn Kim¹, Young Dae Cho³, Chan-Soo Shin³, Won-Kyu Park³, Manny Rivera⁴, Jae Ik Lew⁴, Jung-Hee Lee¹, S. K. Banerjee², Tae-Woo Kim⁵, and Dae-Hyun Kim¹

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Oxidation study on the (100), (110) and (111) surfaces of InAs by ab-initio calculations

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Due to the outstanding electron transport properties of III-V compounds, they became the one of the promising materials for the next-generation semiconductor materials. However, the exposure of a III-V surface to oxygen results in the high density of defect states, and the difficulty of avoiding surface oxidation makes it impossible to use III-V MOSFETs [1]. In order to deal with this problem, changing the surface orientation and the use of compound containing indium was suggested. For example, in GaAs MOSFET, (111)A orientation showed better device performance than the most common (100) orientation [2]. Meanwhile, it has been reported that the electron mobility of InAs is higher than that of GaAs and the MOSFET performances improve significantly when the InAs fraction in the InGaAs channel is increased [3]. In addition, the surface energy of InAs is lower than that of GaAs in most of the surface orientations. Therefore, the theoretical study on InAs surfaces has practical importance and can provide the basis on the further study on InGaAs. In this presentation, the initial stages of oxidation of the multi-orientation surfaces of InAs including (100), (110), (111)A and (111)B were studied by density functional theory (DFT) calculations. The dissociative chemisorption of O₂ has been shown on the transition metal [4], Si(001) [5] and GaAs(001) [6] surfaces. The process on InAs surfaces may be likely similar so we investigated the adsorption of O adatoms on the different surfaces. From the calculated potential energy surface (PES) of O adatoms, the adsorption sites and diffusion energy barriers were identified. Also, the reason of the different device performance depending on the different surface orientations was explained by the construction of detailed bonding structures of the oxidized surfaces.

[1] T. Mimura, K. Odani, N. Yokoyama, Y. Nakayama, and M. Fukuta, IEEE Trans. Electron Devices 25, 573 (1978).

[2] M. Xu, K. Xu, R. Contreras, M. Milojevic, T. Shen, O. Koybasi, Y. Wu, R. Wallace, and P. Ye, IEEE Int. Electron Devices Meet. 865 (2009).

[3] P. Ye, Y. Xuan, Y. Wu, and M. Xu, ECS Trans. 19, 605 (2009).

[4] M. Wheeler, D. Seets, and C. Mullins, J. Chem. Phys. 105, 1572 (1996).

[5] T. Miyake, S. Soeki, H. Kato, T. Nakamura, A. Namiki, H. Kamba, and T. Suzaki, Surf. Sci. 242, 386 (1991).

[6] P. Kruse, J. G. McLean, and A. C. Kummel, J. Chem. Phys. 113, 9224 (2000).